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Analytical method to estimate resin cement diffusion into dentin

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Abstract. This study analyzed the diffusion of two resin luting agents (resin cements) into dentin, with the aim of presenting an analytical method for estimating the thickness of the diffusion zone. Class V cavities were prepared in the buccal and lingual surfaces of molars ($n = 9$). Indirect composite inlays were luted into the cavities with either a self-adhesive or a self-etch resin cement. The teeth were sectioned bucco-lingually and the cement–dentin interface was analyzed by using micro-Raman spectroscopy (MRS) and scanning electron microscopy. Evolution of peak intensities of the Raman bands, collected from the functional groups corresponding to the resin monomer (C–O–C, 1113 cm^{-1}) present in the cements, and the mineral content (P–O, 961 cm^{-1}) in dentin were sigmoid shaped functions. A Boltzmann function (BF) was then fitted to the peaks encountered at 1113 cm^{-1} to estimate the resin cement diffusion into dentin. The BF identified a resin cement–dentin diffusion zone of $1.8 \pm 0.4\ \mu\text{m}$ for the self-adhesive cement and $2.5 \pm 0.3\ \mu\text{m}$ for the self-etch cement. This analysis allowed the authors to estimate the diffusion of the resin cements into the dentin. Fitting the MRS data to the BF contributed to and is relevant for future studies of the adhesive interface. © The Authors. Published by SPIE under a Creative Commons Attribution 3.0 Unported License. Distribution or reproduction of this work in whole or in part requires full attribution of the original publication, including its DOI. [DOI: [10.1117/1.JBO.21.5.055003](https://doi.org/10.1117/1.JBO.21.5.055003)]

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1 Introduction

Adhesion to tooth substrates is fundamentally based on a process whereby the mineral from the superficial region is partially removed and replaced by polymerized resin monomers.¹ This process, known as micromechanical interlocking or hybridization, is believed to induce hybrid layer formation at the adhesive interface.² Establishment of the hybrid layer has been considered important for the bond formed between adhesive materials and the tooth, a significant factor in achieving optimal long-term dental restorations.³

Traditionally, the hybrid layer formed at the interface between adhesive cements and the dentin has been assessed by scanning electron microscopy (SEM), transmission electron microscopy (TEM), or light microscopy.⁴ These morphological imaging techniques have provided substantial insight into the morphological characteristics of resin cement diffusion into dentin that consists of a specific region at the interface, where resin monomers penetrate into the smear layer and demineralized dental hard tissues.⁵ Features such as the extent and uniformity of resin penetration into dentin; appearance; length and continuity of resin tags; quality of hybridization; and hybrid layer thickness have been assessed.^{6–9} Although micromechanical retention resulting from the presence of a hybrid layer and resin tags has been considered the main adhesion mechanism between adhesives and dental substrates, the potential benefit of

additional chemical interactions between functional monomers and the tooth has recently gained attention.^{1,10}

Recently, self-adhesive resin cements have been introduced to the market. Different from the traditional etch-and-rinse or self-etch adhesive materials, self-adhesive cements do not require pretreatment of the tooth surface, and they are applied in a single clinical step, similar to the use of conventional zinc-phosphate or glass ionomer cements.¹¹ Self-adhesive resin cement bonds are attributed to the reaction of acidic monomers present in their formulation with the hydroxyapatite of the dental structure, resulting in chemical and micromechanical retention.¹² However, several studies based on different morphological imaging methods have been unable to demonstrate evidence of dentin decalcification achieved by the self-adhesive resin cements investigated and their infiltration into dentin.^{13–16} Although studies have observed an intimate adaptation of self-adhesive resin cements to dentin, no hybrid layer or resin tag formation could be identified.^{17–19} An established hybridization is typically observed with the use of total-etch or self-etch cements, whereas more superficial physical interactions and possible involvement of chemical reactions between the material and dental hard tissues have been found with self-adhesive cements.¹³

Micromorphological analyses of adhesive interfaces, performed either between the tooth and adhesive systems^{20–22} or between the tooth and resin cements,^{23,24} have made important contributions to the correct use of adhesive materials, with the aim of improved bonding efficiency and durability. More recently, micro-Raman spectroscopy (MRS) has been used as an alternative method to the traditional morphological imaging

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techniques for examining the adhesive interface.^{6,25–28} This method uses a laser beam to scan the area of interest. MRS enables spectra to be collected of the different components present at the adhesive interface. Furthermore, this method identifies the Raman bands associated with the chemical bonds (functional groups), thus allowing researchers to distinguish different materials more precisely. The different Raman bands work like fingerprints for each material analyzed. By observing the different intensities of the spectral bands, variations in the resin cement–dentin diffusion zones between the adhesive materials and dental substrates can be established.^{29,30} Thus, MRS may be an important tool to gain better understanding of the nature of the interface between adhesive cements and dentin.

The purposes of the present study were to analyze the interfaces between two resin luting agents and dentin by means of MRS and to present a mathematical model for estimating the diffusion zone of resin cements into dentin. It was hypothesized that this analysis would allow the authors to measure the diffusion into dentin of both self-etch and self-adhesive resin cements.

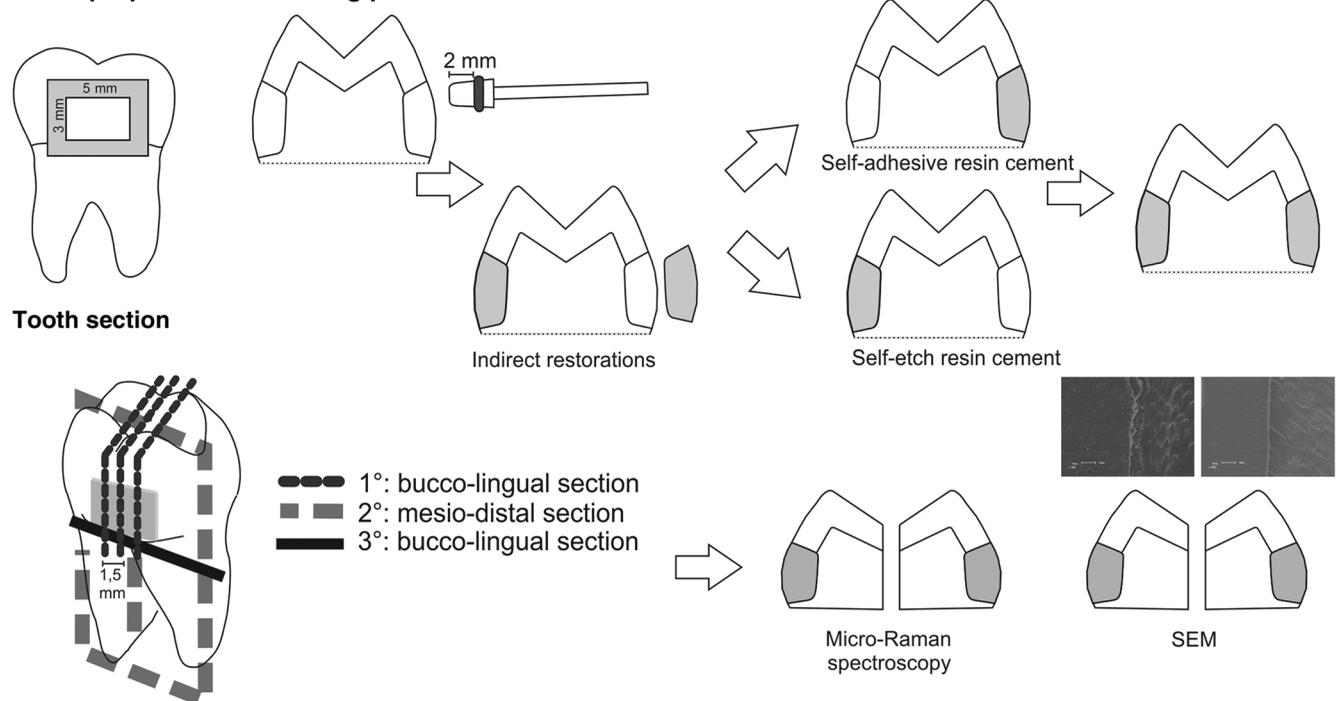
2 Materials and Methods

The study protocol was approved by the Local Ethics Committee (CAAE No. 19425213.2.0000.0104). Nine noncarious human third molars were used in the study ($n = 9$).

2.1 Tooth Preparation and Cementation of Indirect Restorations

Figure 1 illustrates sample preparation and analysis. The teeth were stored in saline solution at 4°C for no longer than six months. Class V cavities (height 3 mm, length 5 mm, depth 2 mm) were prepared at the dentin–enamel junction on the buccal and lingual surfaces of the crown, by using high-speed diamond burs (No. 3131 KG Sorensen, Cotia, SP, Brazil) under water cooling. Due to the bur format selected, the cavities prepared had divergent and expulsive walls. In order to standardize the cavities, stickers with a window measuring 5 mm by 3 mm were placed on the tooth surface, and a rubber stopper was attached to the diamond bur in order to limit the preparation depth. A new diamond bur was used for each tooth. Class V cavity preparations made it possible to evaluate cement adhesion

Tooth preparation and luting procedures



Micro-Raman spectroscopy

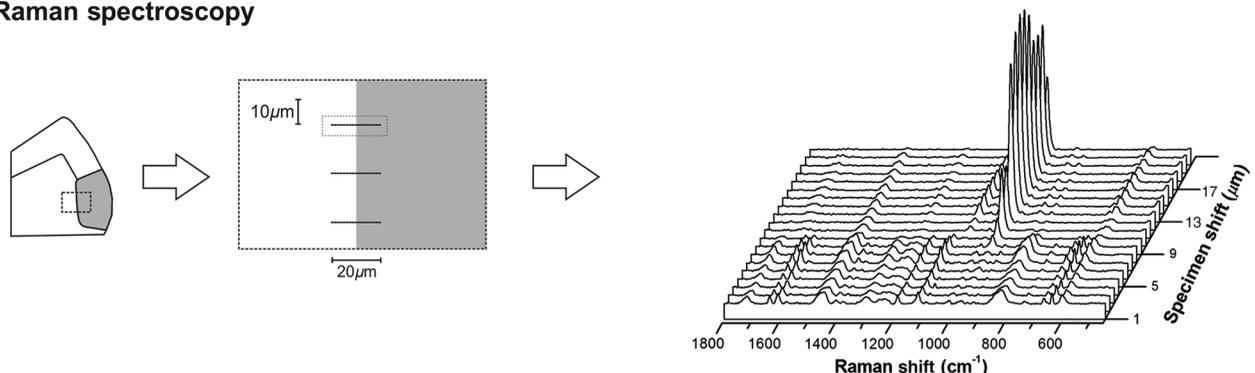


Fig. 1 Schematic drawing illustrating the study setup and specimen preparation.

with axial wall dentin, avoiding extensive wear and allowing paired comparison analysis, by testing different cements in the buccal and lingual surface of the same tooth.

Inlay restorations were fabricated with composite resin (Filtek Z250[®] 3M ESPE, St Paul, Minnesota). The cavities were lubricated with glycerin gel (Insulating gel, Ivoclar-Vivadent, Schaan, Liechtenstein); resin composite was inserted in the cavity and light polymerized with a blue light-emitting diode (LED) light source of 1200 mW/cm² for 20 s. The cavities were rinsed with water and air dried, both for 5 s, before cementation.

Luting procedures were performed in a warm room at 37°C.²⁰ The inlays were luted either with a self-adhesive resin cement (RelyXUnicem 2[®] Clicker, 3M ESPE) or a self-etch resin cement (Multilink[®] Automix, Ivoclar-Vivadent). The luting agent used on the buccal or lingual surface was randomly chosen by draw. In order to allow direct comparisons, both cements were tested in a paired way, i.e., the same tooth received both luting agents. The luting procedures are described in detail in Table 1. The resin cements were light polymerized using the same LED device used to light polymerize the composite inlays.

2.2 Micro-Raman Spectroscopy

After cementation, the specimens were stored in distilled water for 24 h. Then the teeth were fixed with sticky wax and sectioned with a diamond saw (South Bay Technology Inc.,

California) adapted to a low-speed cutting machine (Isomet 1000, Buehler, Lake Bluff, Illinois) under water cooling. Since each tooth had the two analyzed material specimens, the cuts were made in the bucco-lingual direction from the occlusal to the cervical area, across both resin cement–dentin interfaces at the same time, resulting in two slabs 1.5 mm thick (Fig. 1). The inlays were divided into halves, one of which was measured by MRS technique and the other through SEM analysis. To avoid interface sample casting, the specimens were not metallographically polished; to minimize possible debris smeared across the specimen surface, they were submitted to an ultrasonic bath with distilled water for 5 min.

Raman spectra were collected from the interface located at the axial wall of the cavities. The measurements were performed at room temperature with a SENTERRA dispersive Raman microscope (BrukerOptik GmbH, Ettlingen, Germany). The Raman spectra were measured using a micro-Raman system excited by a 785 nm laser source and registered at the spectral range of 450 to 1800 cm⁻¹. Laser power was set at 100 mW and focused on the sample through a 100× objective optical microscope (0.75 N.A.), using the 50 μm confocal hole. The spectral resolution was ~3 to 5 cm⁻¹; detector integration time was 3 s; and the spectrum collected at each point was the average result of 60 consecutive readings. In order to improve signal-to-noise ratio, detector temperature was decreased to -90°C. All spectra were systematically collected under the same conditions. The area chosen for analysis in the first specimen was photographed

Table 1 Application mode of the resin cements tested.

Group	Application mode	Composition
RelyXUnicem 2 [®] Clicker	Tooth cavity: no procedure	Glass powder surface modified acrylic acid methacrylic acid Silane
Self-adhesive	Inlay restoration: no procedure Luting procedure: application of the self-adhesive cement directly on the inlay; placement of the inlay into the cavity under manual pressure for 5 s; removal of excess cement; light activation for 20 s, application of glycerin gel; light activation for additional 20 s.	bulk material 2-propenoic acid, 2-methyl-, 1,1'-[1- (hydroxymethyl)-1,2-ethanediyl] ester, reaction products with 2-hydroxy-1,3- propanediyl dimethacrylate and phosphorus oxide TEDGDMA Silane treated silica Sodium persulfate Oxide glass chemicals (nonfibrous) Tert-butyl peroxy-3,5,5- trimethylhexanoate
Multilink [®]	Tooth cavity: application of primers A and B mixed in the proportion 1:1–30 s on the enamel, 15 s on the dentin; removal of excess with air.	Cement: dimethacrylate, HEMA, barium glass, ytterbium trifluoride, and spheroid mixed oxide
self-etch	Inlay restoration: no procedure. Cementing: application of the self-etch cement on the inlay; placement of the inlay into the cavity under manual pressure for 5 s; removal of excess cement; light activation for 20 s, application of glycerin gel; light activation for further 20 s.	Primer A: aqueous solution of initiators Primer B: HEMA, phosphonic acid, and methacrylate monomers

and served as a reference for locating the interface in the subsequent specimens. Spectra were acquired at positions corresponding to 1- μm intervals across the interface between the resin cement and dentin^{31–38} by using the computer controlled x - y - z stage with a minimum step width of 50 nm. Raman spectra were collected along three line-scans for each slab, starting in the dentin and ending in the cement resin layer, each 20 μm long, with 10 μm space between lines. To avoid specimen dehydration, gauze soaked in distilled water was placed in contact with the dentin to keep it moist during MRS measurement.

2.3 Analysis of the Resin Cements–Dentin Diffusion Zone

The resin cement–dentin diffusion zone between the resin cements and dentin was analyzed by a computer program for interactive scientific graphing and data analysis (Origin, Northampton, Massachusetts). Band intensity peaks at 961 cm^{-1} (ν_1 phosphate symmetric stretch), representative of the phosphate present in dentin, and at $\sim 1113 \text{ cm}^{-1}$ (C–O–C), representative of the carbon chain in the resin monomer (present in both cements), were used to identify the resin cement–dentin diffusion zones. Peaks located at $\sim 1113 \text{ cm}^{-1}$ have previously been used as reference for identifying resin monomers.^{29,30} Nonetheless, in the present study, each spectrum was individually verified to ensure that this peak was present only in the cements tested. The evolution of peak intensities of the Raman bands collected from the functional groups corresponding to the resin monomer present in the cements was sigmoid shaped functions.

A sigmoid function is usually used to determine the rate of growth, starting from low values, passing through its maximum at a point of inflexion, and then reaching a saturation value. The curve looks like a rotated S.³⁹ The inflexion point may be associated with a transition region from the initial to the final stage of the system investigated (Fig. 2). In this study, the experimental data fitted an S-shaped curve, best described by the Boltzmann function.⁴⁰

Figure 2 is representative of the Boltzmann function, which is described by $f(x) = ((A_1 - A_2) / \{1 + \exp[(x - x_0)/dx]\}) + A_2$, where A_1 and A_2 are the initial and final plateau of the peak band intensity at 1113 cm^{-1} , which corresponded to the dentin and

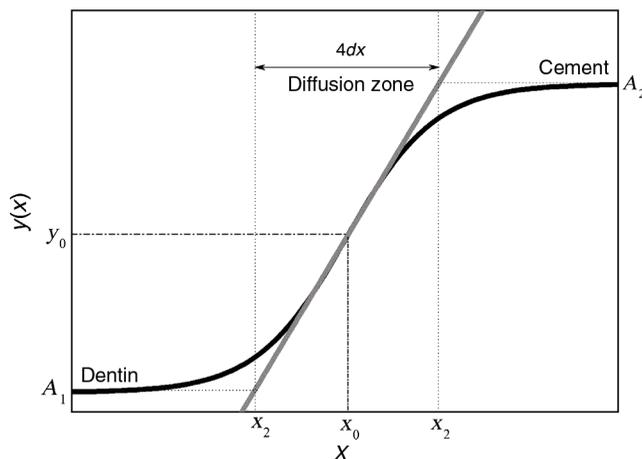


Fig. 2 Boltzmann function used in this study to fit the experimental MRS data; A_1 and A_2 , the initial and final mean values for the dentin and resin cement band intensities; $4dx$, the resin cements–dentin diffusion zone width.

cement region of the sample, respectively. x is the displacement across the interface formed by the adhesive cement and dentin, while x_0 represents the midpoint of the resin cement–dentin diffusion zone. The y axis is the Raman signal intensity. The diffusion zone is the region delimited by a straight-line tangent to the midpoint (x_0, y_0) . The straight tangent line is mathematically described by a linear equation, i.e., $g(x) = ax + b$, where a and b are constants. The constant $= [(A_2 - A_1)/4dx]$ is obtained by the first derivative of the Boltzmann function at $x = x_0$, and the constant $= [(A_2 + A_1)/2] - \{[(A_2 - A_1)/4dx]x_0\}$ is obtained by replacing the values $x = x_0$ and $y = y_0 = [(A_2 + A_1)/2]$ in the $g(x)$ linear function. By replacing a and b values in $g(x)$, it is possible to obtain the distance between x_0 and the two intersection points x_1 and x_2 that correspond to $4dx$, which represents the width of the resin cement–dentin diffusion zones. Thus, by fitting the experimental Raman data collected from the scanned lines to the Boltzmann function, the authors could acquire the dx parameter and determine the width of the resin cement–dentin diffusion zone.

The sigmoidal fitting was performed only for the data collected from the intensity of characteristic peaks associated with cement (1113 cm^{-1}), since the resin cements permeated into the dentin. The characteristic peak associated with dentin (961 cm^{-1}) that represents the mineral component of the tooth did not have a constant distribution across the measurement and was therefore used only as a reference of cement monomer diffusion and was not fitted to the function.

The thickness data were checked for normality using the Shapiro-Wilk test and analyzed by Student's t test for the cement adhesive strategy factor (R-statistics software). Statistical significance was set at $p < 0.05$.

In addition, chemical and physical changes in the spectral region between the dentin and cement were investigated by visual analyses of Raman bands at each individual spectrum for the two groups.

2.4 Scanning Electron Microscopy Analysis

The three specimens selected for SEM (SS-550 Superscan, Shimadzu Biotech, Japan) were dehydrated in increasing series of ethanol for 15 min in each solution of 80 and 90%, and for 30 min in absolute ethanol. The specimens were then sputter-coated with a layer of gold, and photomicrographs (1000 \times magnification) were taken for visual analysis of the cement/dentin interface.

3 Results

From the MRS spectra, the authors identified the resin cement–dentin diffusion zones between the dentin and the resin cements. The characteristic peaks associated with dentin and cement peaks were simultaneously detected for both the self-adhesive [Fig. 3(a)] and the self-etch [Fig. 3(b)] cements. Boltzmann analysis showed significantly lower values ($p < 0.05$) for the resin cement–dentin diffusion zones of self-adhesive cement ($1.8 \pm 0.4 \mu\text{m}$) when compared to self-etch cement ($2.5 \pm 0.3 \mu\text{m}$) [Figs. 4(a) and 4(b)]. Visual analyses of the individual spectra showed that the Raman band correspondent to 600 cm^{-1} (ν_4 phosphate bending mode) was modified in the spectral region between the dentin and cement, thus suggesting the interaction of these materials and representing the resin cement–dentin diffusion zones (Fig. 5).

Scanning electron photomicrography analysis showed differences in the interface between the self-adhesive or

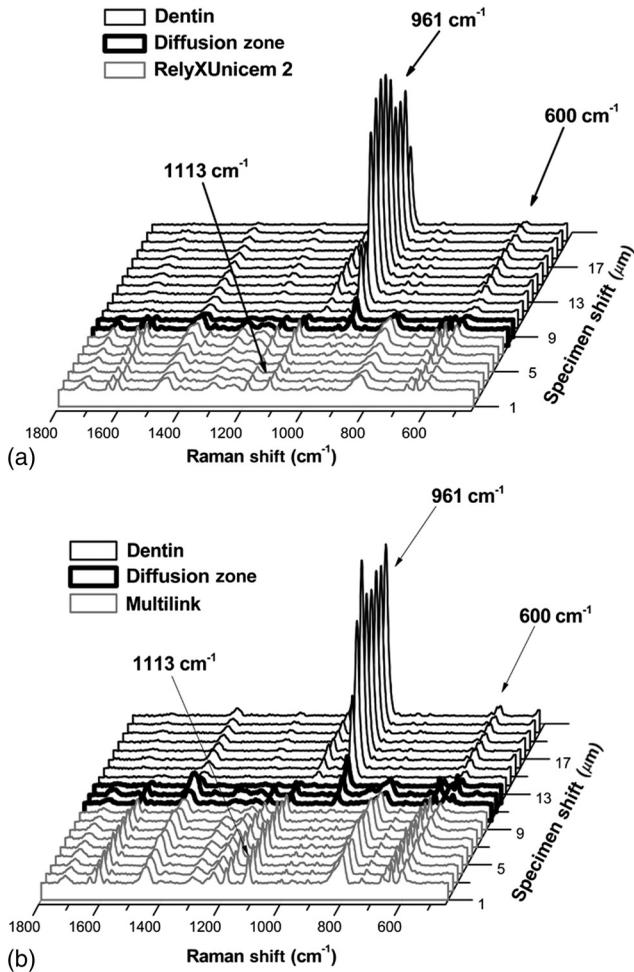


Fig. 3 Representative Raman spectra of lines measured across the dentin–cement interface. The shift scale is represented by a numbered line: (a) RelyXUnicem 2[®] and (b) Multilink[®].

self-etching cements and the dentin. Self-adhesive cement presented an interface that was indistinguishable from dentin [Fig. 6(a)], while the self-etching cement presented a clear transition line with long projections of the material (tags) inside the demineralized dentin [Fig. 6(b)].

4 Discussion

To the best of our knowledge, this is the first study to propose a mathematical model based on Raman spectral readings to estimate the diffusion zone at the interface between resin cements and dentin. The results of the present study demonstrated that there are differences in the thickness of the transition layer between dentin/self-etching cement and dentin/self-adhesive cement. These results were confirmed by the analysis of the photomicrographs that showed that the projections of the self-etching cement were deeper in the dentin, while the tags of the self-adhesive cement showed small penetration (extension) through the dentin. The sigmoidal fitting of the S-shaped curve produced by the MRS data (Fig. 4) resulting from the application of the Boltzmann function to the intensity of the band at 1113 cm⁻¹ allowed the authors to estimate the thickness of the resin cement–dentin diffusion zones. With this new tool, important insight was obtained into the cements investigated.

To date, most studies assessing the interface between adhesive materials and dentin have used SEM, TEM, or light

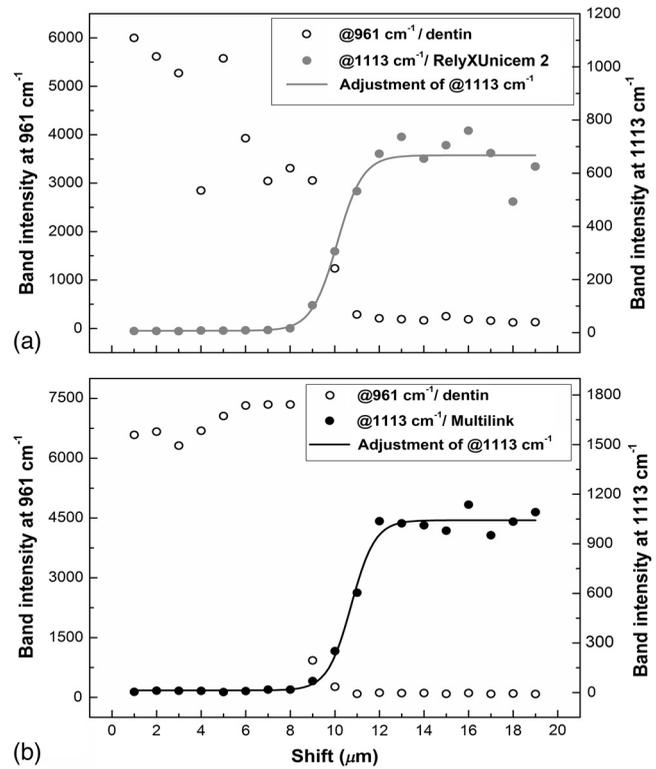


Fig. 4 Ratio of the Raman spectra at 961 cm⁻¹ (ν P–O, dentin) to 1113 cm⁻¹ (ν C–O–C, cement) of one measurement line, and fit to the Boltzmann function at 1113 cm⁻¹. The black spots represent the plot of dentin peaks, which were used only to verify the behavior of dentin in the scanned sample: (a) dentin/RelyXUnicem 2[®] diffusion zone and (b) dentin/Multilink[®] diffusion zone.

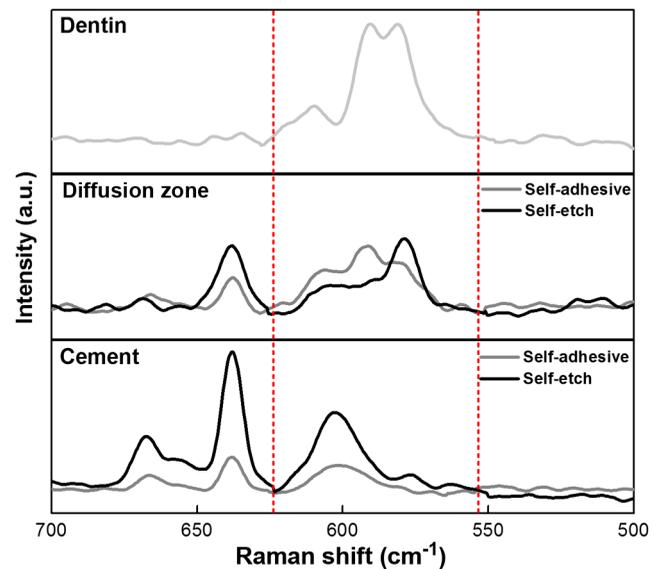


Fig. 5 Illustration of the Raman spectra of a single phosphate band of dentin, the resin cement–dentin diffusion zone, and the resin cements. The spectra delimitation between 627 and 550 cm⁻¹ showed the peak centered at 600 cm⁻¹, demonstrating that at the resin cement–dentin diffusion zone spectra, the phosphate band pattern was altered. Gray line: dentin; dark gray line: RelyXUnicem 2[®]; black line: Multilink[®].

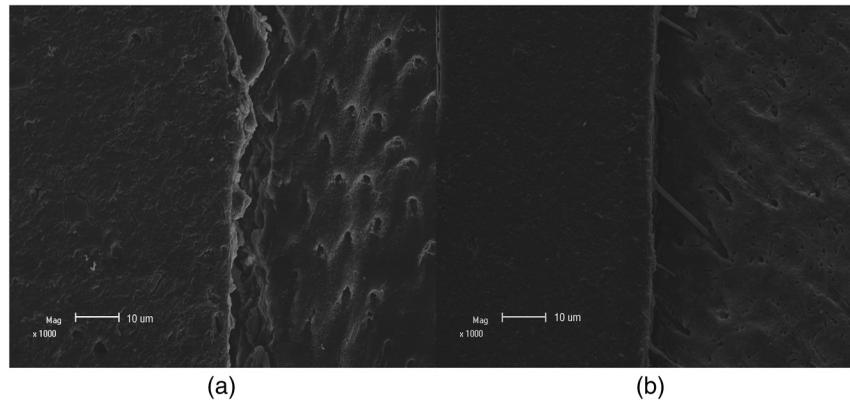


Fig. 6 Representative scanning electron micrographs of the adhesive/cement interface: (a) RelyXUnicem 2[®] and (b) MultiLink[®].

microscopy. Visual analysis of the adhesive interface using these methods can be influenced by the quality of the area being analyzed on the photomicrograph and by sample preparation (sectioning position; level of demineralization; and exposure of dentin tubules). These factors most probably contributed to the variability of the results seen among earlier studies.^{5,12,13,18}

In the search for a more objective procedure to analyze the adhesive interface, Zhang and Wang³⁰ used MRS. Based on the presence of bands representative of dentin and adhesives simultaneously in the same spectrum, the authors were able to locate the adhesive–dentin interface and to estimate the effect of different application modes of the adhesives on hybrid layer formation. In the present study, in a similar manner, a region simultaneously containing traces of P–O (dentin) and C–O–C (resin monomer present in the cements) was detected at the cement–dentin interface for both self-etch and self-adhesive cements.

The results from the present study partly exceed those of previous studies. While morphological imaging techniques have demonstrated a thin hybrid layer for self-etch cements, no such formation was observed for self-adhesive cements, despite the intimate adaptation between the cement and the dental substrate.^{13–19} De Munck et al.,¹⁹ with the use of TEM analysis, observed an irregular interaction zone at the interface between dentin and a self-adhesive cement, ranging from 0 up to 2 μm .

When luting with self-adhesive cements, the dentin tubules were not shown to be infiltrated by the cement.⁵ Moreover, self-adhesive cements were not able to completely dissolve the smear layer.^{5,19} In the present study, we observed an area in which the bands representative of dentin and the self-adhesive cement were simultaneously found in the same spectrum. This clearly demonstrated resin cement diffusion into dentin. A possible explanation is that the smear layer resulting from cavity preparation with diamond burs, usually 1 to 2 μm thick,^{41,42} was infiltrated by resin monomers^{19,43} due to pressure during luting.^{5,19,44} Thus, the MRS results for the self-adhesive cement identified the phosphate present in the smear layer combined with the resin monomer present in the cement.

Mild self-etch adhesives (pH \sim 2) have been reported to form a thin hybrid layer of \sim 1 μm , while an interaction between 1 and 2 μm has been observed with strong (pH < 1) self-etch adhesives.¹ The smear layer formed by this type of adhesive is never rinsed away, but partly dissolved and incorporated into the hybrid layer. In the present study, the slightly higher value found for the diffusion zone of the mild self-etch cement tested ($2.4 \pm 0.4 \mu\text{m}$) may be explained by the fact that the

mineral content present in the dentin and in the smear layer contributed to the measurement of the resin cement diffusion into dentin. Therefore, only part of this value actually refers to the presence of a true hybrid layer.

The results from this study open the possibility for future research, such as assessing the adhesive interface with MRS on smear layer-free dentin surfaces (e.g., fractured dentin) to elucidate this matter. MRS ensures spectral readings that are extremely precise and representative of the materials in the area analyzed. This method does not require surface preparation such as those needed for the morphological imaging techniques, thereby minimizing the possibility of artifacts. However, readings must be verified for possible discrepancies due to the presence of air included at the interface. The main problem with the use of MRS, however, concerns its limited spatial resolution. When 100 \times objective lenses are used, the laser spot is \sim 1 μm , while the adhesive interface may be <1 μm , as shown by ultra-morphological interfacial analysis using TEM.⁵ Since MRS analysis contained hundreds of very high-quality spectra at a spatial resolution of 1 μm ,^{31–38} the difference in phosphate and monomer composition can be determined across the length and thickness of the resin cement diffusion into dentin,^{25,29} with the advantage of the fingerprint characteristics of the Raman spectral bands.

Lenses with 100 \times magnification intensify irregularities in the resin cement–dentin diffusion zones of resin monomers. To overcome this inert feature of the samples, three linear scans were performed on each sample in order to characterize the interface at different positions along the sample. Although the position of the resin cement–dentin diffusion zones differed from one line to another due to irregularities in this region, the chemical analysis with MRS provided a comprehensive representation of the thickness of the resin cement–dentin diffusion zones at the interface. This representation demonstrated a regular pattern for both the self-adhesive ($1.8 \pm 0.4 \mu\text{m}$) and mild self-etch cement ($2.5 \pm 0.3 \mu\text{m}$). The Raman bands obtained with this research are considered fingerprints of specific molecules contained in the specimens. As a result, they provided information on the chemical interactions, expressed by alterations and/or appearance of new peaks.³⁰ The MRS technique enabled the evaluation of the diffusion zone chemical compounds and consequently measured its dimension from its structural composition,^{31–38} while in imaging techniques, the interface is estimated in a qualitative approach.^{6,25–28} Therefore, due to the difference of the characteristics analyzed with these techniques, it is not appropriate to validate

the analytical method proposed in this research through imaging techniques.

In conclusion, MRS was shown to be a simple and reliable method to assess the resin cement diffusion into dentin. The S-shaped curve fitted by the application of the Boltzmann function allowed the authors to estimate the resin cement–dentin diffusion zones. This analysis supplied important information on the extent and nature of the resin cement–dentin diffusion zones. It may benefit future adhesive interface studies of both self-etch and self-adhesive resin cements.

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